Assignment: Logarithmic Applications

For each of the following use the formula: $A = Pe^{rt}$	A = Amount at time t
	P = Original (initial an
	r = Growth rate
	t = time

1. A radioactive substance is decaying. If the initial amount is 10 grams and 6 grams remain after 5 years, then:

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- a) Determine the rate of decay
- b) Determine the amount after 10 years
- 2. It is estimated that 20% of a certain radioactive substance decays in 30 hours.
 - a) Determine the rate of decay
 - b) What is the half-life of this substance?
- 3. A biologist studying a colony of bacteria determines that a certain culture grows exponentially such that the bacteria doubles every 4 days.
 - a) Determine the rate of growth.
 - b) If initially the biologist has 1 200 bacteria present, how many days does it take before 100 000 bacteria are present?
- 4. The half-life of a radioactive carbon 14 is 5 570 years. If 500 milligrams are present today, determine the amount present after 2 500 years.
- 5. A radioactive substance decays from 100 units to 40 units in 25 hours. What is the half-life of this substance?
- 6. Suppose you were able to invest \$1 250 in an account where interest was compounded continuously at a rate of 3%. Find the amount after 4 years.

For each of the following use the	M = Magnitude of earthquake
formula: $M = \log\left(\frac{A}{A_0}\right)$	A = Intensity of vibrations
	A_0 = Intensity of a standard earthquake

7. An earthquake in Vancouver had a magnitude of 6.3 on the Richter scale. An earthquake in Japan had a magnitude of 8.9 on the Richter scale.

How many times more intense was the Japan earthquake than the Vancouver earthquake?

8. A small tremor of magnitude 2.3 is then followed by a stronger one of magnitude 5.3. How much stronger is the second tremor that the first?

For each of the following use the formula: $PH = -LOG[H^+]$	PH = Acid Strength (Power of Hydrogen)
	H ⁺ = Concentration of hydrogen ion

- 9. A beaker of acid has a hydrogen concentration of 3.5 \times 10⁻⁶ mol/L. Calculate the pH of the acid.
- 10.If a beaker of acids has a pH of 4.1, calculate the hydrogen concentration of the acid.
- 11.A beaker of acid has a pH of 4.9, and a second beaker has a pH of 7.6. Determine how many times higher the hydrogen concentration is in the strong acid as compared to the weaker one.

For each of the following use the formula: $A = P \left(1 + \frac{r}{n}\right)^{nt}$

A = Compound amount (new investment value
P = Original amount
r = interest rate (as a decimal)
n = # of Compounding periods
t = time

12.Find the new investment of:

- a) \$2 000 at 8% compounded annually over 15 years
- b) \$5 000 at 12% compounded monthly over 4 years
- c) \$15 000 at 6% compounded semi-annually over 7 years
- 13.You plan to buy a car and have saved \$5 000. The car you want costs \$5 900. How long will you have to invest your money if you can invest at 6.12% compounded quarterly?
- 14. The Robinsons plan to invest money for their newborn son so that he has \$30 000 available for university on his 18th birthday. Assuming a growth rate of 7% compounded semi-annually, how much will the Robinson's need to invest today?
- 15. The amount of a chemical in grams that will dissolve in a solution is given by $C = 8e^{0.3t}$ where t is the temperature in Celsius of the solution. Find t when C = 100 grams.
- 16.Determine how many monthly investments of \$50 would have to be deposited into a savings account that pays 3% annual interest, compounded monthly, for the account's future value to be \$50 000.

 $FV = \frac{R[(1+i)^n - 1]}{i}$ FV = Present Value R = investment payment $i = \frac{annual \text{ interest rate}}{\# \text{ of compounding periods per year}}$ n = # of investments

17.A person borrows \$15 000 to buy a car. The person can afford to pay \$300 a month. The loan will be repaid with equal monthly payments at 6% annual interest, compounded monthly. How many monthly payments will the person make?

$$PV = \frac{R[1 - (1 + i)^{-n}]}{i}$$

$$PV = \text{Present Value}$$

$$R = \text{payment}$$

$$i = \frac{annual interest rate}{\# of compounding periods per year}$$

$$n = \# \text{ of payments}$$

<u>Answers</u>

b) 3.6 g 1. a) -0.1022 2. a) -0.0074 b) 93.2 hours 3. a) 0.1733 b) 26 days 4. r = -0.000124, A = 366.3 mg 5. r = -0.03665, t = 18.9 hours 6. \$1 409.37 7. Vancouver – $10^{6.3}$, Japan – $10^{8.9}$, Japan is 398.1 times as intense 8. Strong – $10^{5.3}$, Small – $10^{2.3}$, Strong tremor is 1 000 times as intense 9. 5.46 10. 7.94 \times 10⁻⁵ mol/L 11.501 12. a) \$6 344.34 b) \$8 061.13 c) \$22 688.85 13. 2.7 years 14. \$8 694.98 15. 8.42°C 16. 502 monthly payments 17. 58 payments